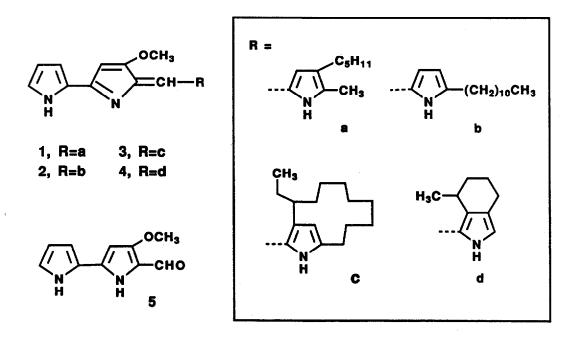
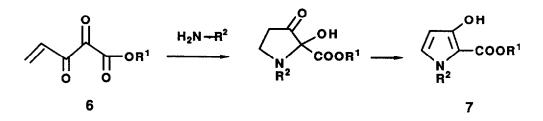
THE CHEMISTRY OF VICINAL TRICARBONYLS A TOTAL SYNTHESIS OF PRODIGIOSIN Harry H. Wasserman* and Louis J. Lombardo Department of Chemistry, Yale University, New Haven, CT 06511 USA

Summary: The addition of primary amines to alkenyl vicinal tricarbonyls leads to 5substituted-3-hydroxypyrroles. This reaction has been employed in a synthesis of the 2formyl-3-methoxy- α, α' -bipyrrole precursor of the prodigiosin family of natural products.

Prodigiosin¹⁻⁵, in the form of the methyl amyl derivative $1,^1$ is the parent member of a group of naturally-occurring pyrryldipyrrylmethenes. These include, among other analogs,⁶ undecylprodigiosin $2,^7$ metacycloprodigiosin $3,^8$ and cycloprodigiosin $4,^9$ which possess potent antibacterial and antifungal activity.⁴ A key intermediate in the biosynthesis⁵ and in the synthesis^{2-4,7-9} of these derivatives is the methoxybipyrrole aldehyde 5 which can be isolated from a mutant strain of *Serratia marcescens*.⁵ We now report a total synthesis of the aldehyde 5 by an application of the chemistry of alkenyl vicinal tricarbonyl compounds.



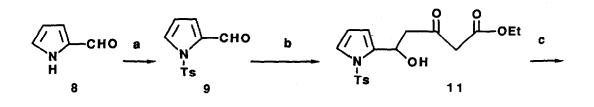
We have recently shown that vinyl vicinal tricarbonyls 6 are converted to 3hydroxypyrrole-2-carboxylates 7 by the double addition of primary amines to the two acceptor sites in the polyfunctional system.^{10,11} When the double bond is conjugated with a pyrrole ring as in 8, this type of reaction may be used for the formation of substituted α, α' bipyrroles, and particularly, for the synthesis of 5. This precursor may be readily transformed by acid-catalyzed condensation with suitable pyrroles to the dipyrromethenes 1,2,3 and 4.

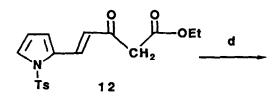


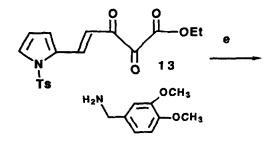
The required unsaturated 1,2,3-tricarbonyl intermediate 12 was formed by the procedure outlined in Scheme 1. Pyrrole-2-carboxaldehyde 8, protected as the 1-p-toluenesulfonyl derivative 9, was treated with the dianion of ethyl acetoacetate 10 in THF at 0°C. The resulting alcohol 11 was dehydrated with gaseous HCl in CHCl₃ at -5°C to yield the enone 12. Our earlier method for forming tricarbonyls from β -keto esters by generating an enamine with DMF acetal followed by cleavage with singlet oxygen¹² was not applicable here due to competing oxidation of the pyrrole ring. We therefore oxidized the active methylene group in 12 according to the method of Sachs¹³ using N,N-dimethyl-p-nitrosoaniline in ethanolic NaOH followed by acid hydrolysis, generating the tricarbonyl derivative 13 as a stable hydrate.

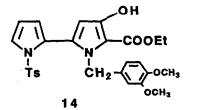
Coupling of 13 with 3,4-dimethyoxybenzylamine yielded a mixture of products from which the intermediate hydroxypyrrolidone carboxylate could be separated and dehydrated (HOAc) to the bipyrrole ester 14. Alternatively, the conversion of 13 to 14 could be accomplished directly on warming 13 with 3,4-dimethoxybenzylamine in glacial acetic acid. This product could then be converted to the methyl ether with dimethyl sulfate. Removal of the dimethoxybenzyl group was accomplished by an adaptation of Evans' oxidative procedure using sulfuric acid/trifluoroacetic acid.¹⁴ Detosylation took place cleanly in ethanolic NaOH providing the known bipyrrole ester 16. As reported earlier, ^{3,4} McFayden-Stevens reduction of 16 yielded the methoxy bipyrrole derivative 5 identical in all respects to the naturally occurring aldehyde formed from the mutant strain of *Serratia*. The further conversion of 5 to prodigiosin and analogues by coupling with methylamyl pyrrole or other substituted pyrroles has been reported in earlier communications.^{2-4,7-9}

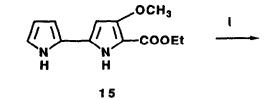
Acknowledgments: This work was supported by NIH Grants GM-07874 and GM-31350

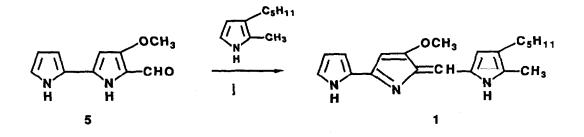












f,g,h

(a) NaH, p-TsCl, THF (93%); (b) $H_2C=C(OLi)CH=C(OLi)OEt 10$, THF (75%); (c) HCl(g), CHCl₃, -5° C (58%); (d) p-Me₂N-C₆H₄-NO, NaOH, EtOH; 1NHCl, THF, 0° C (70%); (e) 3,4-dimethoxybenzyl amine, HOAc(glacial), 60° C (23%); (f) NaH, Me₂SO₄, THF (77%); (g) 5% H₂SO₄, TFA, anisole (62%); (h) NaOH, EtOH, Δ , (65%); (i) 98% H₂NNH₂ (95%); p-TsCl, pyridine (75%); Na₂CO₃, HOCH₂CH₂OH, 170° C (33%); (j) catalytic HBr, (50%).

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